

## Cu-INDUCED CHANGES IN CHLOROPLAST LIPIDS AND PHOTOSYSTEM 2 ACTIVITY IN BARLEY PLANTS

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**Summary.** Twenty-day-old barley plants (*Hordeum vulgare* L., cv. CE9704) grown as a sand culture were treated 10 days with increasing concentrations of Cu 10, 15 and 20 mg/kg sand. Lipid peroxidation, thylakoidal fatty acid content and composition as well as photosynthetic electron transport and chlorophyll fluorescence were studied. It was established that excess Cu enhanced lipid peroxidation processes and decreased total fatty acids content as well as unsaturation level of the thylakoids. These negative effects had a great unfavourable impact on photosynthetic performance which resulted in both lower photosynthetic electron transport activities and efficiency of light utilisation.

**Key words:** barley, copper, lipid peroxidation, ethylene production, photosynthetic electron transport, chlorophyll *a* fluorescence

**Abbreviations:** OEC – oxygen evolving complex; DCPIP – 2,6-dichlorophenolindo-phenol; DPC – 1,5-diphenyl-carbohydrazide; MDA – malondialdehyde; TBA – 2-thiobarbituric acid;  $F_o$  – minimal fluorescence of antennae in dark adapted leaves;  $F_v/F_m$  – photochemical efficiency of PS2;  $q_p$  and  $q_{NP}$

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– photochemical and non-photochemical quenching,  $q_E$  – energy-dependent component of  $q_{NP}$ ;  $F'_v/F'_m$  – PS2 efficiency of energy conversion;  $\phi_e$  – quantum yield of photosynthetic non-cyclic electron transport

## Introduction

Interactions between excess copper and photosynthesis has been studied for a long time (Sandmann and Böger, 1980; Baszynski et al., 1982 Merakchiyska-Nikolova and Yordanov, 1983, Stoinova et al.1999), but the mechanisms of its *in vivo* toxic action are still a matter of debate (Maksymiec, 1997; Shainberg et al., 2001). Noticeably, the functioning of the photosynthetic apparatus of Cu-exposed plants is affected by both direct and indirect mechanisms. The excess Cu causes Rubisco and phosphoenolpyruvat carboxilase (PEPC) inactivation by an interaction with SH-groups (Stiborova et al., 1986; Lidon and Henriques, 1991; Kamenova-Jouhimenko et al, 1999) and bio-production (Kamenova-Youchimenko et al., 2003). In addition, the redox active Cu ion induces membrane lipid peroxidation (Sandmann and Böger, 1980; Weckx and Clijsters, 1996; Shainberg et al., 2001) that further results in serious damages of the thylakoids, namely disintegration of the lamellar system and increased unstacking of thylakoids (Angelov et al., 1993; Lidon et al., 1993 Stoinova and Merakchiyska-Nikolova, 1991). The excess Cu also decreases the photosynthetic pigments affecting both their synthesis and degradation (Merakchiyska-Nikolova and Yordanov, 1983; Lidon and Henriques, 1992a; Vangronsveld and Clijsters, 1994). Also, it alters plant source-sink relationships causing down-regulation of carbon metabolism and primary photochemistry (Ciscato et al., 1997), etc. There are also some data that two wheat *Aegilops ovata* hybrids show good tolerance at seedling stage to high Cu ion concentrations –  $10^{-5}$  M and  $10^{-6}$  M. However, the analysis of the photosynthetic response to excess Cu is rather complicated because of self-regulation processes, the difficulties in distinguishing the primary and the secondary effects as well as the dependence of the plant response to the experimental design used.

In a previous study we found that excess Cu decreased equally both net photosynthetic rate ( $A$ ) and photosynthetic capacity (oxygen evolution under saturating irradiance and  $CO_2 - A_{max}$ ) of barley plants and concluded that  $A_{max}$  inhibition strongly related to mesophyll limitations (Vassilev et al., 2002). In order to clarify the toxic Cu effect on barley photosynthesis we further studied its effects on both chloroplast lipids and PS2 activity.

## Materials and methods

Barley (*Hordeum vulgare* L. cv. CE9704) plants were grown in sand enriched by Hoagland's nutrient solution, in a greenhouse under natural conditions of light,

temperature and humidity as described elsewhere (Vassilev *et al.* 2002). Twenty-day-old plants were exposed for 10 days to Cu treatments by supplying an adequate volume of water Cu solution to final concentrations 0, 10, 15, and 20 mg Cu.kg<sup>-1</sup> sand. At the end of the exposure period plants were harvested and analysed.

### **Chlorophyll fluorescence parameters**

Chlorophyll fluorescence parameters were measured with a PAM 2000 system (H. Walz, Effeltrich, Germany) on leaf discs (from undamaged areas) placed inside the LD2/2 O<sub>2</sub> electrode, under CO<sub>2</sub> saturating conditions, at 25 °C. Measurements of the initial fluorescence from the antennae,  $F_o$ , and photochemical efficiency of PS2,  $F_v/F_m$ , were taken from overnight dark-adapted leaves. The photochemical and non-photochemical quenching,  $q_p$  and  $q_{NP}$ , (Van Kooten and Snel, 1990), the PS2 efficiency of energy conversion,  $F'_v/F'_m$ , (Krupa *et al.*, 1993), and the estimation of quantum yield of photosynthetic non-cyclic electron transport,  $\phi_e$  (Genty *et al.*, 1989) were determined under photosynthetic steady-state conditions, using a photon flux density of 550  $\mu\text{mol.m}^{-2}.\text{s}^{-1}$  as actinic light and 4200  $\mu\text{mol.m}^{-2}.\text{s}^{-1}$  for the saturating flashes (with a duration of 0.8 s). For the measurement of the energy-dependent component of  $q_{NP}$  ( $q_E$ ; Quick *et al.*, 1989), a saturating flash was applied after the recovery of the fast component of the rise observed in  $F'_o$ .

### **Lipid peroxidation and fatty acids determination**

The level of lipid peroxidation products in leaf samples was expressed as 2-thiobarbituric acid-reactive metabolites (mainly malondialdehyde, MDA) according to the method of Heath and Packer (1968). Chloroplast membranes were isolated using 2 g FW leaf tissue according to Droppa *et al.* (1987), with minor modifications, as described by Lidon and Henriques (1992b). The lipid fraction was then extracted from the pellet of chloroplast membranes according to Allen *et al.* (1966), in a mixture of chloroform/methanol/water (1:1:1, v/v/v). After evaporation of the chloroform layer, the dry residue was resuspended in 1 ml of a mixture of ethanol:toluene (1:4, v/v) and stored at -20 °C, until analysis. After saponification with 0.5 M NaOH in methanol, the fatty acids of lipid extracts were methylated with BF<sub>3</sub> (Merck) according to Metcalfe *et al.* (1966) and analyzed by gas liquid chromatography (UNICAM 610 Series Gas Chromatograph, Unicam Ltd., UK). Separation conditions and other details of the method are shown by Campos *et al.* (2003).

### **Ethylene production**

Ethylene production was measured in 500  $\mu\text{l}$  of the obtained chloroplast membranes incubated in 2 ml flasks at a light intensity of 500–600  $\mu\text{mol.m}^{-2}.\text{s}^{-1}$ , provided by a Björkman lamp. After two hours of incubation a 1 ml gas sample was withdrawn from

the headspace gas of the incubating flask using a gas-tight syringe. Ethylene concentration in this gas sample was assayed by a Pye Unicam Series 204 gas chromatograph equipped with a Porapak Q column and a flame ionization detector (FID). Nitrogen, at a flow rate of 30 ml.min<sup>-1</sup>, was the carrier gas. The temperatures were set to 90°C for the oven, room temperature for the injection port, and 150°C for the detector. Ethylene was identified and quantified by comparison with the peak area from the gas samples containing a known concentration (29 µmol.mol<sup>-1</sup>) of ethylene standard.

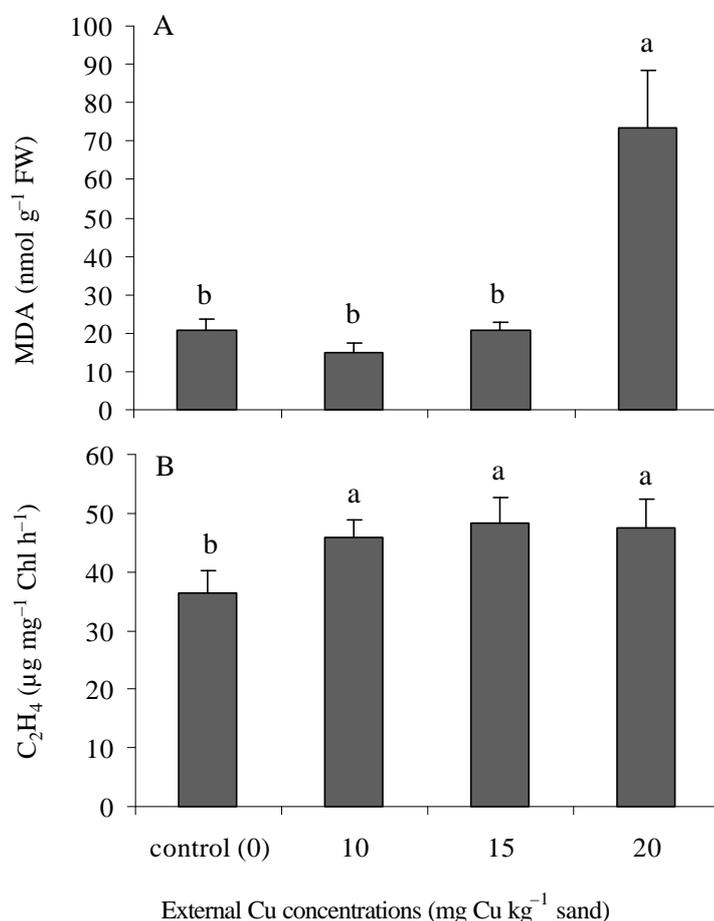
### Photosynthetic electron transport

Determination of photosynthetic activities coupled to PS2 were measured with a Clark-type oxygen electrode (LW2, Hansatech, Kings Lynn, UK), using the chloroplast membranes obtained as described by Droppa *et al.* (1987), with minor modifications, described by Lidon and Henriques (1992b). The electron transport rates were determined according to Droppa *et al.* (1987) in 1 ml of reaction mixture containing 100–150 µg chl, at 25°C and with PPFR of 3000 µmol.m<sup>-2</sup>.s<sup>-1</sup>, given by a Bjorkman lamp (Hansatech).

Statistical analysis was performed using a one way ANOVA (for P<0.05).

## Results

We measured the peroxidation status of Cu-exposed barley plants by two different methods: by the commonly used test based on thiobarbituric acid reactive metabolites (mainly MDA), and by ethylene production associated with extracted thylakoids. The level of lipid peroxidation products in the leaves of Cu-exposed barley plants, measured as thiobarbituric acid reactive metabolites, was similar to that of control plants at the first two treatments (10 and 15 mg Cu.kg<sup>-1</sup>), but sharply increased (almost 3-fold) at the highest treatment 20 mg Cu.kg<sup>-1</sup> (Fig. 1A). On the other hand, the ethylene production associated with the extracted thylakoids was significantly higher in all Cu treatments (Fig. 1B). Since ethylene production associated with thylakoid membrane degradation and mediated by oxy radicals and H<sub>2</sub>O<sub>2</sub> is a final product of acyl lipid peroxidation (Lidon and Henriques, 1993), the obtained results showed that excess Cu triggered lipid peroxidation processes in the chloroplasts. Cu treatment clearly resulted in a reduction of total fatty acids (TFA) content of the chloroplast membranes. The decrease in total fatty acids (TFA) was about 57% for 20 mg Cu.kg<sup>-1</sup> (Table 1). The strongest decrease was detected in the content of linolenic (C18:3) and *trans* hexadecenoic (C16:1*t*) fatty acids, being lower as compared with the control plants by 65% and 68%, respectively. Cu treatments slightly changed fatty acids percentage, but the level of fatty acid unsaturation tended to decrease as indicated by the double bond index (DBI) although it was not significantly different (Table 1).



**Fig. 1.** Leaf lipid peroxidation (A) and ethylene production (B) of photosynthetic membranes isolated from barley plants grown for 10 days in Cu-contaminated sand. Means  $\pm$ SE (n = 3). Within the same parameter, values followed by the same letters are not significantly different ( $P \leq 0.05$ ).

The photosynthetic performance of Cu-exposed barley plants was evaluated by the activity of photosynthetic electron transport and by chlorophyll fluorescence measurements. The results obtained by both techniques showed that excess Cu inhibited photosynthetic functioning. The activity of photosynthetic electron transport involving PS2 was significantly lowered (Fig. 2). The inhibiting effect of Cu increased gradually with the Cu concentration applied. After a 10 days exposure to 20 mg Cu.kg<sup>-1</sup>, the decrease in PS2 activity with OEC was about 45% and without OEC – 49%. Chlorophyll fluorescence response of Cu-exposed plants also showed some negative changes (Table 2). Significant reductions of  $q_p$  and  $\phi_e$  values as well as increases in  $q_{NP}$  and  $q_E$  values have been detected in the last two Cu treatments (15 and 20 mg Cu.kg<sup>-1</sup>).

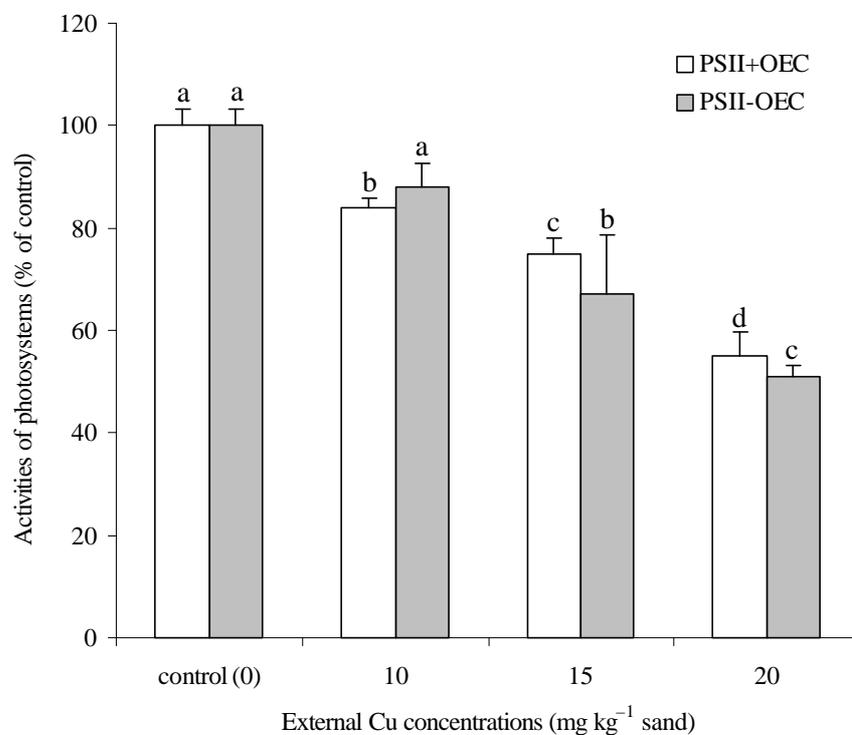
**Table 1.** Total (TFA) fatty acids content (mg.g DW<sup>-1</sup>), composition (mol %) and unsaturation (DBI) of thylakoid membrane lipids isolated from barley plants grown for 10 days in Cu-contaminated sand. The determined fatty acids were palmitic (16:0), *cis* and *trans* hexadecenoic (16:1), stearic (18:0), oleic (18:1), linoleic (18:2) and linolenic (18:3). The values represent the mean of triplicates. Within each row, data followed with the same letters are not statistically different ( $P \leq 0.05$ ).

Parameter	control (0)	10 mg Cu kg <sup>-1</sup>	15 mg Cu kg <sup>-1</sup>	20 mg Cu kg <sup>-1</sup>
	(mg.g <sup>-1</sup> DW)			
TFA	164.2 a	154.3 a	138.2 a	69.9 b
C18:3	36.8 a	35.7 a	31.6 a	12.7 b
C16:1 <sub>t</sub>	2.5 a	1.8 a	1.2 b	0.8 b
	(mol %)			
< C16:0	10.19 a	12.98 a	11.89 a	15.88 a
C16:0	18.36 ab	16.11 b	17.71 ab	22.37 a
C16:1 <sub>c</sub>	3.96 ab	3.92 b	4.97 ab	5.51 a
C16:1 <sub>t</sub>	3.74 a	2.92 ab	2.15 b	2.69 a
C18:0	0.64 b	0.57 b	0.69 ab	0.96 a
C18:1	0.73 ab	0.58 b	0.65 b	1.11 a
C18:2	6.22 a	5.02 a	4.94 a	6.64 a
C18:3	56.13 ab	57.88 a	56.95 ab	45.53 a
DBI	6.68 a	6.53 a	6.24 a	4.16 a

Double bond index (DBI)=[(C16:1<sub>c</sub>+C16:1<sub>t</sub>+C18:1+2xC18:2+3xC18:3) / (<C16:0+C16:0+C18:0)]

## Discussion

The obtained results generally confirmed our suggestion that Cu-induced decrease of net photosynthetic rate (A) in barley plants was related to mesophyll limitations (Vassilev et al., 2002). We identified as a reason for these constraints the ability of Cu to act as an efficient generator of toxic oxygen species that might initiate lipid peroxidation processes (Sandmann and Böger, 1980). Both tests applied gave us an evidence for enhanced lipid peroxidation in Cu-exposed barley plants. The established higher level of thiobarbituric acid reactive metabolites in leaves of the plants at 20 mg Cu.kg<sup>-1</sup> treatment probably are due to the oxidizing properties of Cu itself (Weckx and Clijsters, 1996) as well as Cu-provoked stimulation of lipoxygenase activity (Gora and Clijsters, 1989). Cu-induced ethylene production in the thylakoids corresponds with the finding of Lidon et al. (1993) in Cu-exposed rice plants. These authors suggested that it is due to lipid peroxidation processes along with the concurrent formation of linolenate hydroperoxides. Further, excess of Cu can reduce these peroxides via Fenton-type reaction to alkoxy radical, splitting an ethyl radical which in a reaction with Cu is able to produce ethylene (Lidon, 1999).



**Fig. 2.** Rates of photosynthetic electron transport between H<sub>2</sub>O and 2,6-dichlorophenol-indo-phenol (DCPIP) (PSII+ oxygen evolving complex - OEC) and 1,5-diphenyl-carbohydrazide (DPC) and DCPIP (PSII-OEC) in thylakoid membranes isolated from leaves of barley plants grown for 10 days in Cu-contaminated sand. Control values (representing 100%) were 63.6 and 52.4  $\mu\text{mol O}_2 \text{ mg.chl}^{-1}.\text{h}^{-1}$  for PSII+OEC and PSII-OEC, respectively. Values are the mean  $\pm$ SE (n=3). Within the same parameter, values followed by the same letters are not significantly different ( $P \leq 0.05$ ).

**Table 2.** Selected chlorophyll fluorescence parameters and quenching analysis coefficients in leaves of barley plants grown for 10 days in Cu-contaminated sand. Within each column, data followed with the same letters are not statistically different ( $P \leq 0.05$ ).

Parameter	$F_o$	$F_v/F_m$	$F'_v/F'_m$	$q_P$	$q_{NP}$	$q_E$	$\phi_\epsilon$
control (0)	44.8 a	0.797 a	0.488 a	0.497 a	0.782 a	0.472 a	0.242 a
SE	0.3	0.004	0.019	0.012	0.011	0.033	0.013
10 mg Cu kg <sup>-1</sup>	44 a	0.801 a	0.473 a	0.489 a	0.813 a	0.5 a	0.233 a
SE	1.1	0.007	0.014	0.034	0.011	0.016	0.023
15 mg Cu kg <sup>-1</sup>	42.4 a	0.811 a	0.508 a	0.426 b	0.798 a	0.565 b	0.216 b
SE	0.7	0.001	0.008	0.017	0.005	0.005	0.01
20 mg Cu kg <sup>-1</sup>	41.5 a	0.808 a	0.473 a	0.442 b	0.824 b	0.556 b	0.208 b
SE	1.1	0.001	0.014	0.033	0.008	0.036	0.011

Drastically reduced TFA content of the barley thylakoids at 20 mg Cu.kg<sup>-1</sup> treatment as a result of the enhanced lipid peroxidation may further lead to increased membrane leakiness and permeability. The decreased membrane unsaturation level (expressed in a lower DBI value) may be explained by the strong decrease of the linolenic acid (C18:3) in the same treatment. An evidence for the altered fatty acid desaturation process in Cu-exposed plants has been already reported (Lanaras et al., 1993; Quariti et al., 1997). The established lower content of *trans* hexadecenoic (C16:1*t*) may decrease the oligomerization of the light-harvesting complex 2 (LHC2). According to Krupa and Baszynski (1995) the level of C16:1*t* fatty acid is positively correlated with LHC2 oligomerization, because of its specifically binding in *sn*-2 position in the chloroplastic phosphatidylglycerol. Thus, the decreased C16:1*t* content may diminish the content of LHC2 oligomer that means less efficient energy collection and distribution between photosystems.

The inhibiting Cu effect on photosynthetic electron transport in barley plants corresponds with the results of Baszynski et al. (1982) and Lidon and Henriques (1991) found in spinach and rice plants, respectively. The lower  $f_e$  observed in plants at 15 and 20 mg Cu.kg<sup>-1</sup> treatment, clearly indicated lower efficiency of light utilisation (Table 2). Obviously, it was due to a decrease in  $q_p$  as  $F'_v/F'_m$ , a measure of PSII photochemical efficiency under steady-state light conditions (Krupa et al., 1993), was not depressed considerably. The smaller fraction of the open PS2 reaction centers ( $q_p$ ) together with a  $q_{NP}$  and  $q_E$  tendency to increase supports the suggestion of Maksymec (1997) for feedback control due to reduced demand for ATP and NADPH. In this way, Cu-induced alterations in primary carbon metabolism (Stiborova et al., 1986; Lidon and Henriques, 1991; Angelov et al., 1993) may lead to down-regulation of PS2 activity. To some extent this may be also due to an alteration of source-sink relationship with a consequently diminished requirement for products of photosynthesis as was suggested by Ciscato et al. (1997).

In conclusion, the 10 days exposure of barley plants to excess Cu induced serious disturbances of the chloroplast membranes and PS2 activity. The observed increase of the thiobarbituric acid reactive metabolites as well as the ethylene production associated with the thylakoids gave some evidences for Cu-enhanced lipid peroxidation processes. These negative effects had a great impact on photosynthetic performance presented by both lower electron transport activities and efficiency of light utilisation.

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